ACID CATALYZED INTRAMOLECULAR CONJUGATE ADDITION AS A ROUTE TO trans HYDRINDANE SYSTEMS

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Abstract A stereocontrolled acid catalyzed intramolecular conjugate addition and its application in the enantioselective synthesis of trans hydrindanes are described

In this communication we wish to report an acid catalyzed stereocontrolled cyclization² $(\underline{1} \rightarrow \underline{2})$ which might be applicable in the construction of chiral precursors for steroid synthesis

The starting material for these studies was the readily available cis-3-methyl-4-carboxy-cyclohexene ($\underline{3}$) 3 The ester was reduced with LiAlH₄ and the resulting alcohol $\underline{4}$ was transformed to the enone $\underline{1}$ by a three step sequence—Ozonolysis (CH₂Cl₂, -78°C, Me₂S workup) of the olefin $\underline{4}$ gave the lactol $\underline{5}$ which was treated with (C₆H₅)₃P=CHCOCH₃ (benzene, r t) to afford the enone $\underline{6}$ Compound $\underline{6}$ on stirring with methanol (CH₂Cl₂, Py⁺Ts, r t) furnished the enone $\underline{1}^5$ in 51% yield from ester $\underline{3}$.

On treatment with catalytic TsOH in $\mathrm{CH_2Cl_2}$, enone $\underline{1}$ was converted to the bicyclic product $\underline{2}$ in 70% yield. The fact that the cyclization leading to $\underline{2}$ was very stereoselective with respect to the newly formed carbon-carbon bond was proved by its hydrolysis (aq acetone, HCl) to lactol $\underline{7}$ followed by oxidation (PCC, $\mathrm{CH_2Cl_2}$) to the lactone $\underline{8}^6$ (72% overall yield), IR (CHCl₃) 1760, 1720 cm⁻¹, NMR (CDCl₃) δ 2 17, 2 16 (singlets, 3H), 1 32, 1 18 (singlets, 3H, ratio 1 10)

The proof that the major isomer corresponds to structure 2 follows from its conversion to

$$3 R = COOMe$$

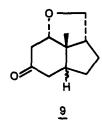
 $4 R = CH2OH$

$$5 R = 0, R^1 = R^2 = H$$

$$\underline{\underline{3}}$$
 R = COOMe
 $\underline{\underline{4}}$ R = CH₂OH $\underline{\underline{6}}$ R = CHCOMe, R¹ = R² = H
 $\underline{\underline{16}}$ R = COOMenthyl $\underline{\underline{1}}$ R = CHCOMe, R¹ = Me, R² = H

 $\frac{2}{7} R^1 = Me, R^2 = H$ $\frac{7}{7} R^1 = R^2 = H$

8



$$R^1$$
 R^2

 $\underline{10} R^1 = H, R^2, = OCOMe$

 $11 R^1 H, R^2 = OCOMe$

 $\frac{-1}{12}$ R¹ = H, R² = OH $\frac{13}{13}$ R¹ = H, R² = SePh(o-NO₂)

14 R = CH₂ 15 R = 0

 $17 R^1 = H, R^2 = CH_2OH$

the known trans hydrindanone derivative $\underline{15}$. ^2a The lactol $\underline{7}$ on heating with KOH in aqueous methanol gave the tricyclic product $\underline{9}$ (60% from $\underline{2}$, two epimers by V P C, $\underline{4}$ ratio 1 10). Compound $\underline{9}$ was transformed into the enone $\underline{10}$ (72%) by heating with acetic anhydride and TsOH in benzene Catalytic hydrogenation (EtOAc, Pd-C) of $\underline{10}$ and saponification of the resulting acetate $\underline{11}$ provided the alcohol $\underline{12}$ (88%) which on oxidative elimination $\underline{7}$ of the corresponding selenide $\underline{13}$ 8 provided the olefin $\underline{14}$ (40% from $\underline{12}$) Ozonolysis (CH₂Cl₂, Me₂S workup) of $\underline{14}$ gave the trans diketone $\underline{15}$ ^{2a} as the major product $\underline{9}$

The predominance of the desired anti relationship of the angular methyl and hydrogen at the newly formed carbon-carbon bond in $\underline{2}$ can be rationalized by considering the transition states \underline{A} and \underline{B} 10

The steric congestion in transition state \underline{B} leading to cis isomer of $\underline{15}^{11}$ makes it less favorable than the alternative transition state \underline{A} which favors the trans ring junction.

With the assumption that cyclization $\underline{1} \to \underline{2}$ will proceed without loss of optical activity, we decided to prepare the key intermediate $\underline{1}$ in optically active form. Diels Alder reaction of trans piperylene and R(-)menthyl acrylate (CH₂Cl₂, EtAlCl₂, -20°C) provided the ester $\underline{16}^{13}$ (70%) in 43% optical yield. The transformation of compound $\underline{16}$ into the alcohol $\underline{12}$ proceeded as described for $\underline{3} \to \underline{12}$ Compound $\underline{12}$ had in fact been formed from $\underline{16}$ without loss of optical activity Making use of the chiral acrylate derived from (-)- β -pinene¹⁶ the optical yield of compound $\underline{1}$ could be improved to 55-60%

In conclusion, we have shown that the acid catalyzed intramolecular conjugate addition of systems such as $\underline{1}$ can be used to prepare optically pure 17 intermediates with controlled stereochemistry. Compounds such as $\underline{10}$ can serve as valuable intermediates in the synthesis of chiral steriods. 18

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References and Notes

- 1 Address correspondence to this author at The Squibb Institute for Medical Research, P O Box 4000, Princeton, NJ 08540
- 2 For related base catalyzed cyclizations see (a) Stork, G, Shiner, CS, Winkler, JD J Am. Chem Soc 1982, 104, 310, (b) Stork, G, Atwal, K, Tet Lett 1982, 2073, Stork, G, Winkler, JD, Saccomano, NS Tet Lett 1983, 465

- 3 Inukai, T, Kojima, T J J. Org Chem., 1967, 32, 869. EtAlCl₂ in CH₂Cl₂ gives better yield (58% to alcohol 4) with 94% endo selectivity 3
- 4. V P C. analysis was performed on SE-30 fused silica glass capillary column (25m x 0 24mm) using Packard Gas Chromatography Model 433.
- 5. The structure of compound <u>1</u> was proved by its hydrolysis (aq. acetone, HC1) followed by oxidation (PCC, CH₂Cl₂) of lactol <u>6</u> to the five-membered lactone which absorbs at 1747 cm⁻¹ in I.R
- 6 V P C analysis 4 shows it to be a mixture, ratio 1 10, minor isomer elutes faster
- 7 Sharpless, K. B., Young, M W J. Org. Chem., 1975, 40, 947
- 8. Grieco, P , Gilman, S , Nishizawa, M. J Org Chem , 1976, 41, 1485.
- 9. The minor product (0.4% at this stage) shows the same retention time 3 as an authentic sample of cis-isomer 2a
- 10 The evidence for the intermediacy of \underline{A} and \underline{B} is circumstantial at this stage. The cyclization $(\underline{1} \rightarrow \underline{2})$ was unsuccessful in MeOH. The unstable intermediate $(\underline{A},\underline{B})$, prepared by dehydration (MsCl/pyridine) of $\underline{6}$, refused to cyclize cleanly. Additionally, if the ketoaldehyde $\underline{17}$ (open form of $\underline{1}$) is an intermediate (which could have been trapped with MeOH under acid catalysis) one would not expect high stereoselectivity during cyclization [1]
- It is possible that cis product was formed from the open form ($\underline{17}$) of $\underline{1}$ as acid catalyzed cyclization of such systems shows very little stereoselectivity. Small amount (\sim 5%) of a byproduct $\underline{18}$ originating from $\underline{17}$ was actually isolated
- 12 Unpublished results from these laboratories
- 13 94% endo by V.P C analysis 4 of alcohol 4
- 14 The optical purity of compound $\underline{16}$ (and $\underline{12}$) was determined by preparing the Mosher ester 15 of alcohol 4 (and 12)
- 15 Dale, J A , Dull, D L., Mosher, H. S J Org. Chem, 1969, 34, 2543
- 16. Oppolzer, W et al Helv Chim. Acta, 1981, 64, 2802
- 17. Chiral cis-3-methyl-4-carboxy-1-cyclohexene (3, R = COOH) of course can be made by optical resolution of the (±) adduct: Monroe, J. D., Ph.D., thesis, Yale University, 1974 Since the sequence 4 + 12 proceeds without loss of optical activity, one can therefore obtain compound 10 in optically pure form.
- 18 Stork, G , Winkler, J. D , Shiner, C S J. Am Chem Soc 1982, 104, 3767 (Received in USA 3 June 1983)